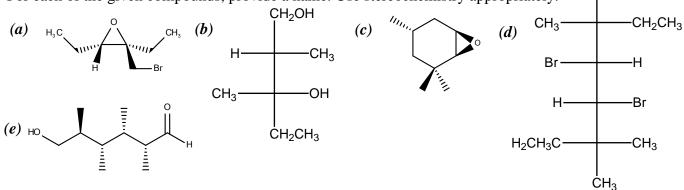
CH₂CH₃

Organic Chemistry I Practice Set #9 (Chapters 7 – Carey)

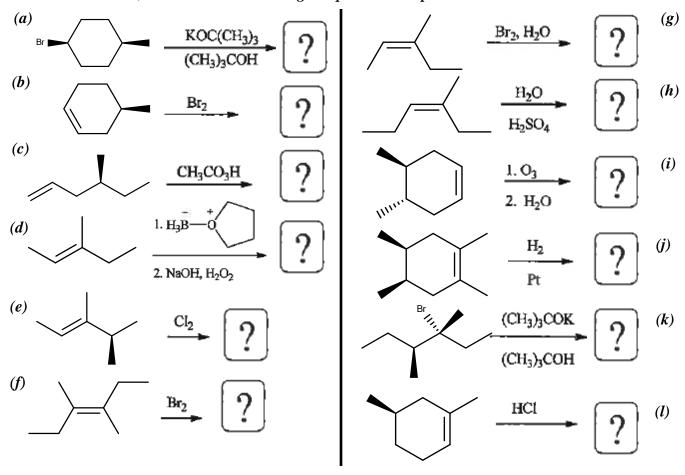
1) For each of the given compounds, provide a name. Use stereochemistry appropriately.



- 2) For this problem, consider the structural formula given in *problem 1e*.
 - (a) Does it represent a D or L sugar?
 - (b) Represents: (i) aldopentose; (ii) aldohexose; (iii) ketopentose; (iv) ketohexose?
 - (c) Give a Fischer projection of it in standard orientation for a sugar.
 - (d) Give a Fischer projection of a compound that is an epimer of compound 1e and a D sugar
 - (e) Give a Fischer projection of a compound that is an epimer of compound 1e and a L sugar
 - (f) Give a Fischer projection of a compound that is a diastereomer, but not an epimer of compound 1e and a D sugar
 - (g) Same as above, make it a L sugar
 - (h) Give a Fischer projection of the compound that is the enantiomer of compound 1e.
- 3) Describe the relationship of each pair of molecules: (a) same molecule (b) constitutional isomers (c) diastereomers (d) enantiomers

Adapted from practice handouts created by Dr. EF Hilinski of Florida State University

4) Fill in what is missing. Either give all of the missing reagents to complete the reaction or give a structural formula for the *major organic product(s)*. Show stereoisomers properly if necessary. If no reaction occurs, write *N.R. Each starting compound is the pure stereoisomer shown*.



IMPORTANT NOTE for above problem! If there is a mixture of isomers for the major product, put both. Write "equal" if they are formed in equal amounts and "unequal" if they are formed in unequal amounts.

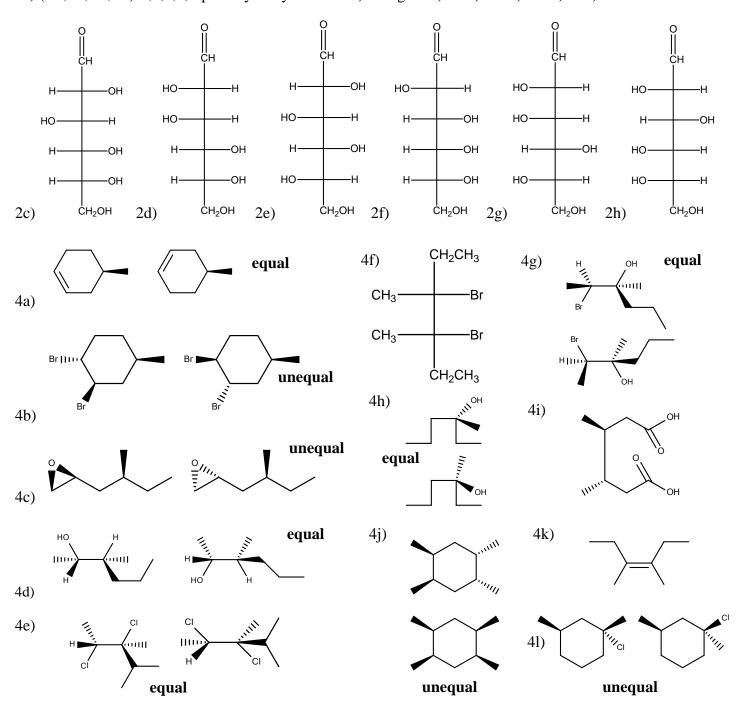
5) Using arrows to show the flow of electrons, write a stepwise mechanism for each of the following reactions. Show clearly how the final stereoisomeric products are formed.

(a)
$$+ Br_{2} + H_{2}O \Rightarrow OH \longrightarrow CH_{3} CH_{3} \longrightarrow OH \longrightarrow Br$$

$$CH_{2}CH_{3} CH_{3} \longrightarrow OH \longrightarrow CH_{3} CH_{3} \longrightarrow OH \longrightarrow H \longrightarrow CH_{3} CH_{3} \longrightarrow OH \longrightarrow H \longrightarrow OH \longrightarrow CH_{3} CH_{3} \longrightarrow OH \longrightarrow H \longrightarrow OH \longrightarrow CH_{3} CH_{3} \longrightarrow OH \longrightarrow CH_{3} \longrightarrow$$

Organic Chemistry I Answers to Practice Set #9 (Chapters 7 – Carey)

- 1a) (2R,3S)-2-(bromomethyl)-2,3-diethyloxirane 1b) (2S,3R)-2,3-dimethylpentane-1,3-diol
- 1c) (2S,3R,5S)-2,3-epoxy-1,1,5-trimethylcyclohexane 1d) (4R,5R)-4,5-dibromo-3-ethyl-3,6,6-trimethylcyclohexane
- 1e) (2R,3S,4R,5R)-2,3,4,5,6-pentahydroxyhexanal 2a) D sugar 2b) ii 3a) iv 3b) iii 3c) i 3d) iv



Organic Chemistry I Answers to Practice Set #9 (Chapters 7 - Carey) iв **t**\\!>-_{t1}\\ O'fl'4.{S **fM**, (i $\setminus \{L$ $\backslash \backslash 3C - + -$ 3) Proton-transfer to create alcohol Both B and B' undergo I will only show one B'3 Final Water attacks each I in the rive of A with equal probability because both ('s are identically substituted

<u>B'</u>

_B